

- II. 0.520 gm. of substance dried at 110° C., treated with hydrofluoric and sulphuric acids, required for the determination of protoxide of iron 6.5 c.c. of permanganate of potash = 0.0395 gm. of protoxide of iron (1 c.c. of permanganate of potash = 0.005846 gm. of protoxide of iron).
- III. 1.0252 grms. of substance dried at 110° C. gave 0.0127 gm. of potash and, by difference, 0.0288 gm. of soda.

Silica,	.	.	.	.	.	.	.	.	.	50.56
Titanic acid,	.	.	.	.	.	.	.	.	.	0.80
Alumina,	.	.	.	.	.	.	.	.	.	10.80
Peroxide of iron,	.	.	.	.	.	.	.	.	.	4.95
Protoxide of iron,	.	.	.	.	.	.	.	.	.	7.59
Protoxide of manganese,	.	.	.	.	.	.	.	.	.	0.14
Lime,	.	.	.	.	.	.	.	.	.	9.85
Magnesia,	.	.	.	.	.	.	.	.	.	9.27
Potash,	.	.	.	.	.	.	.	.	.	1.24
Soda,	.	.	.	.	.	.	.	.	.	2.81
Water,	.	.	.	.	.	.	.	.	.	1.70
										<hr/> 98.71

80. PUMICE.—Station 241.

Lat. 35° 41' N., long. 157° 42' E., 2300 fathoms (Renard).

- I. 1.2725 grms. of substance dried at 110° C., fused with the carbonates of soda and potash, gave 0.7755 gm. of silica, 0.2032 gm. of alumina, 0.1155 gm. of peroxide of iron, 0.0371 gm. of lime, 0.0629 gm. of loss on ignition, 0.0493 gm. of pyrophosphate of magnesia = 0.0178 gm. of magnesia.
- II. 1.0515 grms. of substance dried at 110° C., treated with hydrofluoric and sulphuric acids, gave 0.0707 gm. of the chlorides of soda and potash, 0.0914 gm. of chloroplatinate of potash = 0.0169 gm. of potash and, by difference, 0.0246 gm. of soda.

Silica,	.	.	.	.	.	.	.	.	.	60.95
Alumina,	.	.	.	.	.	.	.	.	.	15.97
Peroxide of iron,	.	.	.	.	.	.	.	.	.	9.08
Lime,	.	.	.	.	.	.	.	.	.	2.92
Magnesia,	.	.	.	.	.	.	.	.	.	1.40
Potash,	.	.	.	.	.	.	.	.	.	1.61
Soda,	.	.	.	.	.	.	.	.	.	2.34
Loss on ignition,	.	.	.	.	.	.	.	.	.	4.95
Manganese,	.	.	.	.	.	.	.	.	.	large trace
										<hr/> 99.22

81. BASIC VOLCANIC GLASS.—Station 285.

Lat. 32° 36' S., long. 137° 43' W., 2375 fathoms (Renard).

- I. 0.8463 gm. of substance, fused with the carbonates of soda and potash, gave 0.4510 gm. of silica, 0.1258 gm. of alumina, 0.0991 gm. of ferric oxide, 0.0788 gm. of lime, 0.1834 gm. of pyrophosphate of magnesia = 0.06619 gm. of magnesia.
- II. 0.5448 gm. of substance, treated with hydrofluoric and sulphuric acids, required for oxidation 6.6 c.c. permanganate of potash solution (1 c.c. permanganate of potash solution = 0.0058463 gm. of ferrous oxide) = 0.03859 gm. of ferrous oxide.
- III. 1.5701 grms. of substance gave 0.0265 gm. of water (loss on ignition).
- IV. 1.5235 grms. of substance, treated with hydrofluoric and sulphuric acids, gave 0.08159 gm. of the chlorides of potash and soda, 0.0243 gm. of chloroplatinate of potash, corresponding to 0.0074 gm. of chloride of potash = 0.0047 gm. of potash, and 0.0741 gm. of chloride of soda = 0.0392 gm. of soda.