

No. 18.—Station XXI. March 13th. Lat.  $18^{\circ} 54'$  N., Long.  $61^{\circ} 28'$  W. Depth, 3025 fathoms. Bottom temperature,  $1^{\circ} 3$  C. Chemical composition:

Loss on ignition after drying at $230^{\circ}$ F.....	5·92														
Portion soluble in hydrochloric acid = 50·42.	<table border="0"> <tr> <td>Alumina.....</td> <td>7·04</td> </tr> <tr> <td>Ferric oxide.....</td> <td>12·25</td> </tr> <tr> <td>Calcium phosphate.....</td> <td>Small traces</td> </tr> <tr> <td>Calcium sulphate.....</td> <td>0·51</td> </tr> <tr> <td>Calcium carbonate.....</td> <td>2·44</td> </tr> <tr> <td>Magnesium carbonate.....</td> <td>3·48</td> </tr> <tr> <td>Silica.....</td> <td>24·70</td> </tr> </table>	Alumina.....	7·04	Ferric oxide.....	12·25	Calcium phosphate.....	Small traces	Calcium sulphate.....	0·51	Calcium carbonate.....	2·44	Magnesium carbonate.....	3·48	Silica.....	24·70
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Portion insoluble in hydrochloric acid = 43·66.	<table border="0"> <tr> <td>Alumina.....</td> <td>5·51</td> </tr> <tr> <td>Ferric oxide.....</td> <td>6·73</td> </tr> <tr> <td>Lime.....</td> <td>0·81</td> </tr> <tr> <td>Magnesia.....</td> <td>0·41</td> </tr> <tr> <td>Silica.....</td> <td>30·20</td> </tr> </table>	Alumina.....	5·51	Ferric oxide.....	6·73	Lime.....	0·81	Magnesia.....	0·41	Silica.....	30·20				
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A red clay, containing much amorphous clayey matter, with iron peroxide; many fragments of sanidine, augite, olivine, hornblende, and magnetite; many of the mineral particles much larger than those at Station XX.—A few fragments of the tests of *Globigerina*.

No. 19.—Station XXII. March 14th. Lat.  $18^{\circ} 40'$  N., Long.  $62^{\circ} 56'$  W. Depth, 1420 fathoms. Bottom temperature,  $3^{\circ} 0$  C. Chemical composition:

Loss on ignition after drying at $230^{\circ}$ F.....	3·80													
Portion soluble in hydrochloric acid = 92·75.	<table border="0"> <tr> <td>Alumina.....</td> <td rowspan="2">{ 4·42</td> </tr> <tr> <td>Ferric oxide.....</td> </tr> <tr> <td>Calcium phosphate.....</td> <td>2·41</td> </tr> <tr> <td>Calcium sulphate.....</td> <td>0·41</td> </tr> <tr> <td>Calcium carbonate.....</td> <td>80·69</td> </tr> <tr> <td>Magnesium carbonate.....</td> <td>0·68</td> </tr> <tr> <td>Silica.....</td> <td>4·14</td> </tr> </table>	Alumina.....	{ 4·42	Ferric oxide.....	Calcium phosphate.....	2·41	Calcium sulphate.....	0·41	Calcium carbonate.....	80·69	Magnesium carbonate.....	0·68	Silica.....	4·14
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Silica.....	4·14													
Portion insoluble in hydrochloric acid = 3·45.	<table border="0"> <tr> <td>Insoluble residue, principally alumina and ferric oxide, with silica .....</td> <td>3·45</td> </tr> </table>	Insoluble residue, principally alumina and ferric oxide, with silica .....	3·45											
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A globigerina ooze, containing many pelagic foraminifera of the genera *Globigerina*, *Orbulina*, *Pulvinulina*, *Pullenia*, and *Sphaeroidina*; many shells of pteropods and heteropods; a few coccoliths and rhabdoliths; otolites of fishes, and spines of echini; a few siliceous spicules.—Amorphous mineral matter and particles of quartz, feldspar, hornblende, and magnetite.